Electronic Structure of Nd_{1-x}Sr_xFeO₃ Probed by Soft-X-Ray Spectroscopy

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The electronic structure of Nd_{1-x}Sr_xFeO₃ (x=0, 0.1 and 0.4) has been studied by soft-X-ray spectroscopy. The Nd_{1-x}Sr_xFeO₃ has the mixed valence states of Fe³⁺ and Fe²⁺. The intensity of hole-induced state at the bottom of the conduction band increases with increasing Sr concentration. The valence band is mainly composed of the O2*p* state hybridized with the Fe 3*d* state. The resonant photoemission spectra of Nd_{0.9}Sr_{0.1}FeO₃ exhibit the e_g -subband of Fe 3*d* state at near E_F , which is closely related to the electrical conductivity at room temperature.

1 Introduction

Sr-doped NdFeO₃ (Nd_{1-x}Sr_xFeO₃) with perovskite structure is expected as a *p*-type thermoelectric material due to its high chemical stability in air. The electrical conductivity for the oxygen gas partial pressure exhibits hole-ion mixed conduction, which needs for electrode of solid state oxide fuel cell (SOFC), at high temperature region [1]. The mixed conduction is useful for the performance of SOFC. However, details of structural and electrical properties have not been clarified far.

In this study, the authors have prepared the Nd_{1-x}Sr_xFeO₃ bulk ceramics and probed their electronic structures by X-ray absorption spectroscopy (XAS) and high-resolution photoemission spectroscopy (PES). The Nd_{1-x}Sr_xFeO₃ has hole and oxygen vacancies in the crystal lattice by Sr substitution. In this paper, the authors discuss about the valence state of Fe, oxygen vacancies and electronic structures of the valence band for Nd_{1-x}Sr_xFeO₃ (*x*=0, 0.1, 0.2, and 0.4) in comparison with electrical conductivity.

2 Experiment

Nd_{1-x}Sr_xFeO₃ (*x*=0, 0.1, 0.2, and 0.4) bulk ceramics were prepared by the solid-state reaction method. Nd₂O₃, SrCO₃ and Fe₂O₃ powders were mixed using a wet ball mill. The mixture was pressed into a disk shape at 14 kN and sintered for 24 h at 1300°C. The crystallization of the prepared target was confirmed using X-ray diffraction (XRD) using CuK α .

The electronic structure was confirmed by photoemission spectroscopy (PES) and X-ray absorption spectroscopy (XAS), which was installed at the undulator beamline BL-2A in Photon Factory, KEK. The XAS spectra were measured by total-electron-yield mode. The kinetic energy of photoelectron for PES was measured using Gammadata-Scienta SES-2002. The energy resolutions of PES and XAS were 100 meV and 60 meV, respectively, at hv=700 eV. The energy axis was calibrated by measuring Au film evaporated onto the manipulator in all measurements.

3 Results and Discussion

Table 1 shows peak positions of (110), (112) and (202) for Sr concentration estimated from XRD patterns. The (110) and (112) peaks shifts to higher diffraction angle

side and (202) peak shifts to lower diffraction angle side with increasing Sr concentration. This result indicates that the lattice constant of a-axis increases and those of b- and c-axes decrease with Sr concentration.

Table 1 Peak positions of (110), (112) and (202) for the XRD patterns

	(110)	(112)	(202)
<i>x</i> =0	22.78 °	32.50 °	40.36 °
x=0.1	22.86 °	32.54 °	40.32 °
x=0.2	22.88 °	32.58 °	40.26 °
x=0.4	22.90 °	32.64 °	40.20 °



Tempareture (K)

Fig.1 Temperature dependence of electrical resistivity of $Nd_{1-x}Sr_xFeO_3$ (*x*=0, 0.1, 0.2, and 0.4).

Figure 1 shows the temperature dependence of electrical resistivity of $Nd_{1-x}Sr_xFeO_3$. The NdFeO₃ exhibits insulator-like behavior. The resistivities of *x*=0.1 and 0.2 decrease with decreasing temperature between 400 and 1000°C. However, those increase below 400°C. On the other hand, the resistivity of *x*=0.4 decreases with decreasing temperature, which corresponds to the metallic behavior. The resistivity of Nd_{1-x}Sr_xFeO₃ is lower than that of La_{1-x}Sr_xFeO₃.

Figure 2 shows the Fe 2*p* XAS spectra of Nd_{1-x}Sr_xFeO₃. Each spectrum consists of two parts derived from the spin-orbit split of L_3 (2*p*_{3/2}) and L_2 (2*p*_{1/2}) states. They are



Fig. 2 O 1s XAS spectra of $Nd_{1-x}Sr_xFeO_3$ (*x*=0, 0.1, 0.2, and 0.4).



Fig. 3 Fe 2p XAS spectra of Nd_{1-x}Sr_xFeO₃ (*x*=0, 0.1, 0.2, and 0.4).

further split into the t_{2g} and e_g states due to the octahedral ligand field. The crystal-field splitting (10Dq) corresponding to the energy separation between t_{2g} and e_g states is 1.5 eV. The spectral shape and peak position are similar to those of La_{1-x}Sr_xFeO₃, which was calculated assuming a high-spin $[t_{2g}]^3[e_g]^2$ ground state [2]. The intensity of t_{2g} -subband increases with Sr concentration, indicating that the doped Sr ions are doped as hole.

Figure 3 shows the O 1s-XAS spectra of Nd_{1-x}Sr_xFeO₃. The O 1s XAS of Fe oxide corresponds to the transition from O 1s core level to unoccupied O 2p state hybridized with Fe 3d state. The conduction band is manly composed of Fe 3d state. Hole or oxygen vacancies induced state (*h*) is observed at the bottom of the conduction band. The intensity of the structure increases with Sr concentration.

Figure 4 shows the resonant-PES spectra of $Nd_{0.9}Sr_{0.1}FeO_3$ measured at various excitation energies. The off-resonance spectrum reflects the O 2p states without Fe 3*d* state. The intensity of on.1 spectrum corresponding to t_{2g} -resonance is enhanced in the valence band region, indicating the hybridization effect between



Fig. 4 Fe 2p resonant PES spectra of Nd_{0.9}Sr_{0.1}FeO₃ measured at various excitation energies of Fig. 3.

O 2*p* state and Fe 3*d* (t_{2g}) state. The on.2 spectrum corresponding to e_g -resonance exhibits strong resonance effect at peak α at near the Fermi level (E_F). This result indicates that the e_g -subband of Fe 3*d* state is closely related to the electrical conductivity of Nd_{0.9}Sr_{0.1}FeO₃.

4 Summary

We have studied the electrical resistivities and electronic structures of $Nd_{1-x}Sr_xFeO_3$. The lattice constant of *a*-axis increases and those of *b*- and *c*-axes decrease with Sr concentration. The Fe has the mixed valence states of Fe²⁺ and Fe³⁺. The hole-induced state with Sr doping is observed at the bottom of the conduction band. The valence band is mainly composed of t_{2g} - and e_{g} subbands of Fe 3*d* states hybridized with O 2*p* states. The electrical conductivity of Nd_{0.9}Sr_{0.1}FeO₃ is achieved by the e_{g} -subbnad at E_{F} .

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Research Achievements

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