## BL-27B/2015G154 EXAFS on tetravalent uranium cations in molten lithium fluoride

Haruaki MATSUURA<sup>\*1</sup>, Takafumi UCHIYAMA<sup>1</sup>, Yukiko OKADA<sup>1</sup>, Atsushi NEZU<sup>2</sup>, Hiroshi AKATSUKA<sup>2</sup>, Mathieu GIBILARO<sup>3</sup>, Didier ZANGHI<sup>4</sup>, Catherine BESSADA<sup>4</sup>

<sup>1</sup>Dept.Nucl. Safety Eng., Tokyo City University, Tamazutsumi, Setagaya-ku, Tokyo 158-8557, Japan <sup>2</sup>Res. Lab. for Nucl. Reactors, Tokyo Tech., Ookayama, Meguro-ku, Tokyo, 152-8550, Japan <sup>3</sup>Chem.Eng.Lab., CNRS-UMR 5503, Paul Sabatier University, 118 route de Narbonne, 31062 Toulouse, France

<sup>4</sup>CEMHTI, CNRS, 1D avenue de la recherche scientifique, 45071 Orléans cedex 2, France

## **Introduction**

A molten salt reactor concept contains still fascinating idea from the point of view of its self-sustainability constructed by a closed fuel cycle. In order to develop an on-line recycle process of molten salt fuels, one of promising technologies is an electrochemical separation of actinides (An) from lanthanides (Ln). To find the best electrolysis condition to improve the efficiency of the pyrochemical process, systematic investigations of the correlation between structures of molten An  $(Ln)F_n$  and their physico-chemical properties, such as electrochemical behavior are useful. In this report, molten UF<sub>4</sub>-LiF mixtures are focused as the structural investigation by EXAFS.

## **Experimental**

All UF<sub>4</sub> mixture samples were prepared in France and imported to Japan. To avoid the contamination by oxygen, premixing treatment was abandoned. Mixtures of UF<sub>4</sub> and LiF were weighted then, they were mixed with boron nitride powder, and pressed into pellets in 7 mm diameter and 1 mm thickness. The mixing weight ratio of UF<sub>4</sub> to BN was ca. 1: 2.5. To prevent chemical reaction of sample and contamination of UF<sub>4</sub> by atmosphere during performing high temperature EXAFS measurements, these pellets were installed in a double barrier cell, i.e., the 1<sup>st</sup> barrier is made with pyrolytic boron nitride and the 2<sup>nd</sup> barrier is made with boron nitride ceramics. The electric furnace chamber was filled with He gas under ca. 30 kPa. Uranium  $L_{m}$ -edge EXAFS spectra have been collected with fixed time scan method by the X-ray from a double Si (111) crystals monochromator in transmission mode. EXAFS data were analysed by using the WinXAS ver.3.1 and 3<sup>rd</sup> and 4<sup>th</sup> cumulants were introduced for the curve fitting analyses of EXAFS data at molten phase due to apperance of their large anharmonic effect in some cases.

## **Results and discussion**

By the stepwise-controlled temperature program, the spectra at molten phase were well identified in each

sample. The concentration dependence of extracted EXAFS oscillations at molten phase is shown in Fig. 1.



Fig. 1 EXAFS oscillations of LiF-UF<sub>4</sub> ( $x_{UF4} = 0.05, 0.10, 0.15, 0.20, 0.30$ ) at molten phase.

Temperature of each spectra is varied since its melting temperature is different according to the phase diagram. On the contrary to the case of ThF<sub>4</sub> mixtures, the 1<sup>st</sup> neighbor contribution which is corresponding to  $U^{4+} - F^-$  correlation doesn't seem to be well modified. It is not necessarily to tell that there is no modification of network-like structures depending upon the concentration of UF<sub>4</sub> in LiF. A molecular dynamics simulation using well established inter-ionic potential parameters would be very helpful to evalute the microscopic structure.

This study has been carried out under collaboration research between French institutes and Japanese universities.

\*hmatuura@tcu.ac.jp