

## Pressure-Induced Liquid-to-Liquid Crossover in GeI<sub>4</sub>

Kazuhiro FUCHIZAKI<sup>1,\*</sup>, Hironori NISHIMURA<sup>1</sup>, Takaki HASE<sup>2</sup> and Hiroyuki SAITOH<sup>3</sup>

<sup>1</sup>Department of Physics, Ehime University, Matsuyama, 790-8577, Japan

<sup>2</sup>Technical Division, ANSYS JAPAN, Tokyo, 160-0023, Japan

<sup>3</sup>Quantum Beam Science Research Directorate, National Institutes for Quantum and Radiological Science and Technology, Hyogo 679-5148, Japan

### 1 Introduction

We could confirm in SnI<sub>4</sub> the existence of the liquid—liquid critical point (LLCP) at around 1.5 GPa and 1000 K [1] and of the temperature of maximum density in the region between 1100 K and 1400 K just below 1 GPa [2]. SnI<sub>4</sub> thus allows us to enter the “no-man’s land”, which cannot be accessed for bulk water

It should be emphasized that a melting maximum [3] is not necessarily a condition for a liquid—liquid transition (LLT) to occur. The existence of a (seemingly) singular point on a melting curve may rather imply occurrence of an LLT [4]. Indeed, the melting curve of SnI<sub>4</sub> does not exhibit an apparent maximum but has a breakpoint at ~1.5 GPa [4], which may be the triple point among the two liquid phases and the crystalline phase (CP) [5].

The melting curve of a molecular crystalline GeI<sub>4</sub> exhibits quite a similar shape (see Figure 1, in which the melting points are marked by open circles). The breakpoint is shifted to ~3 GPa [6] reflecting a slight reduction of molecular size. The aspect suggests us to examine a possible LLT of liquid GeI<sub>4</sub> in a pressure—temperature region close to the breakpoint.

### 2 Experiment

To this end, the energy-dispersive synchrotron x-ray diffraction measurements were carried out employing MAX-80, a cubic-type multianvil press, installed in the beamline NE5C at KEK-AR. A new type of two-stage compression method that was described previously in full by [6] was employed. The compression was performed using a set of anvils with a truncation edge length of 6 mm to cover the relevant pressure range. NaCl was used as the pressure standard. The distributions of scattered x-ray intensities at various scattering angles, 3 through 20°, were combined to construct a normalized structure factor, which was then converted to the reduced radial distribution function (RRDF) using the method by [7].

### 3 Results and Discussion

The characteristic feature of the RRDF was found to be completely shared with that found for SnI<sub>4</sub> [7]; the intramolecular I—I distance shrinks on compression whereas the intramolecular Ge—I distance *expands*. This latter unusual feature enables us to judge the occurrence of the LLT from the low-pressure liquid state (Liq-II) to the high-pressure liquid state (Liq-I). The distribution of structure measurements at which Liq-II (blue circles) and Liq-I (pink circles) were identified is depicted in Figure 1.

Unlike the aspect observed for SnI<sub>4</sub>, the transition between the two liquids is sluggish; transient states (purple circles) were indeed in the region near the breakpoint. The LLCP for GeI<sub>4</sub> might be located within the CP field below the melting curve, making the transition merely a crossover. The more complete discussion was given in [8].

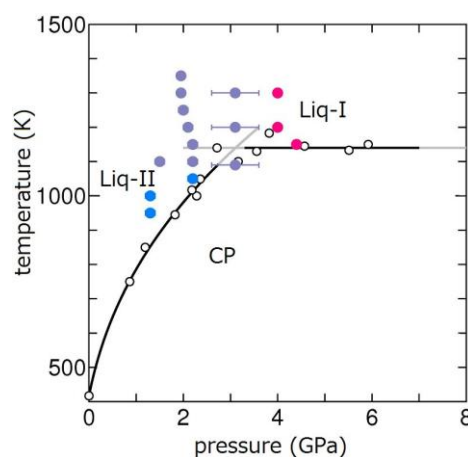


Fig. 1: Two liquid states of GeI<sub>4</sub> and their coexistence.

### Acknowledgement

This work was supported by JSPS KAKENHI (Grant Nos. 26400398 and 17K05581). We thank the Geodynamics Research Center, Ehime University for the use of facilities.

### References

- [1] K. Fuchizaki *et al.*, talk presented at StatPhys 26, Lyon, July 2016; to be published.
- [2] K. Fuchizaki *et al.*, talk presented at EHPRG 2016, Bayreuth, September 2016; in preparation.
- [3] E. Rapoport, *J. Chem. Phys.* **46**, 2891 (1967).
- [4] K. Fuchizaki, *J. Chem. Phys.* **139**, 244503 (2013).
- [5] K. Fuchizaki *et al.*, *J. Chem. Phys.* **135**, 091101 (2011).
- [6] K. Fuchizaki and N. Hamaya, *J. Phys. Soc. Jpn.* **83**, 074603 (2014).
- [7] K. Fuchizaki *et al.*, *J. Chem. Phys.* **130**, 121101 (2009).
- [8] K. Fuchizaki *et al.*, *J. Phys.: Condens. Matter* **30**, 045401 (2018).

\* fuchizak@phys.sci.ehime-u.ac.jp