

Stability of Ce³⁺ valence state and lattice expansion of CeO₂ induced by high temperature annealing in vacuum

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1 Introduction

In our previous papers, we have shown that energetic ion irradiation produces oxygen vacancies in CeO₂ [1]. Oxygen vacancies in CeO₂ cause the change of the valence state of Ce atoms near the oxygen vacancies from 4+ to 3+ as a result of charge balance [2] and the lattice expansion due to the mutual repulsion of Ce cation atoms [3]. We have also shown that the existence of Ce³⁺ valence state induces the ferromagnetism in the CeO₂ specimens [3,4]. It is well known, however that oxygen vacancies can be produced in CeO₂ more easily under the reduction process by the high temperature annealing in vacuum. In the present experiment, the stability of Ce³⁺ state and the lattice expansion induced by the high temperature annealing in vacuum was investigated by using the x-ray photoelectron spectroscopy (XPS) at the BL-27A beamline of KEK-PF and a conventional x-ray diffraction method (XRD).

2 Experiment

CeO₂ pellet was synthesized by sintering CeO₂ powders at 1873K for 12 hours in air. It was annealed at 1273 K for 1 hour in vacuum, and just after the annealing in vacuum, it was stored in an evacuated glass tube until the XPS spectra were measured. Just before the specimen was set in an XPS chamber, it was taken out of the glass tube. The specimen was left in the atmosphere only for 15 minutes before evacuating the XPS chamber. After the XPS measurement, the specimen was left in the atmosphere for 50 minutes and then the XPS spectra were measured again. The energy of x-ray for the XPS measurements was 2.2 keV. For the XRD measurement with Cu K α x-ray, a CeO₂ thin film, which was deposited on an Al₂O₃ substrate by using a RF magnetron sputtering, was used.

3 Results and Discussion

The XPS spectrum (a) in Fig. 1, which was measured for the CeO₂ specimen without high temperature annealing in vacuum, shows typical Ce-3d peaks for Ce⁴⁺ valence state. In the figure, the reference spectrum for the Ce³⁺ valence state is also plotted (spectrum (d)) [2]. The spectrum (b) is for the CeO₂ specimen which was exposed to the atmosphere for 15 minutes before the XPS measurement. XPS peaks for Ce³⁺ valence state can be observed as well as those for Ce⁴⁺ state in the spectrum. The Ce³⁺ state and the oxygen vacancies in the specimen, therefore, survive the exposure to the atmosphere for 15 minutes. As can be seen in the spectrum (c), however, Ce³⁺ peaks completely

disappear, and only Ce⁴⁺ peaks can be observed after the exposure to the atmosphere for 50 minutes. The present experimental result clearly shows that the Ce³⁺ valence state and oxygen vacancies produced by the high temperature annealing in vacuum is very unstable, and oxygen atoms in the atmosphere re-oxidize the CeO₂ specimen including oxygen vacancies in a very short time.

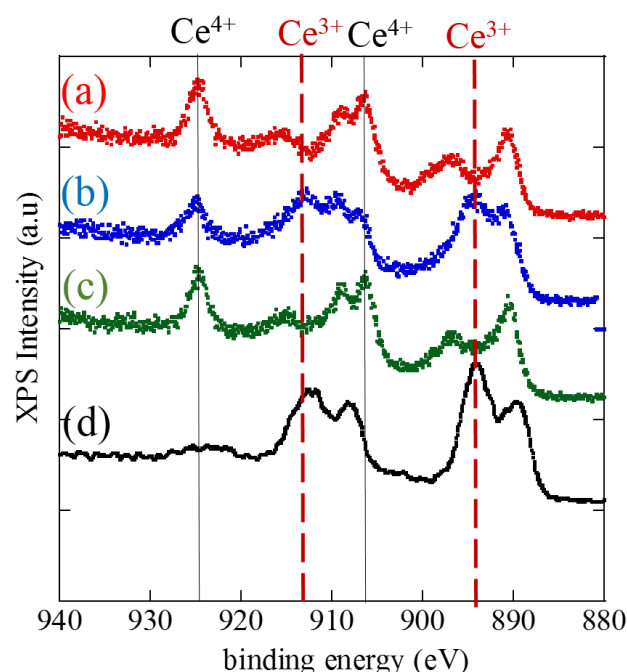


Fig. 1: (a) XPS spectrum for CeO₂ specimen without high temperature annealing, (b) XPS spectrum measured after exposure to atmosphere for 15 minutes, (c) XPS spectrum measured after exposure to atmosphere for 50 minutes, and (d) reference spectrum for Ce³⁺ valence state.

The instability of oxygen vacancies produced by high temperature annealing in vacuum has also been confirmed by the XRD measurement [5]. Fig. 2 shows the time-dependence of XRD spectra for CeO₂ thin foil annealed at 1273 K in vacuum. The peak of the XRD spectrum for the CeO₂ thin foil, which was exposed to the atmosphere for 21 minutes before the XRD measurement, shifts to lower angles, meaning that the lattice expansion occurs by the oxygen vacancy production. With increasing the exposure

time to the atmosphere, the XRD peaks gradually shift to higher angles and after the two-day exposure, the peak position is about the same as for the spectrum of the specimen before the high temperature annealing. Therefore, the exposure to the atmosphere for 2 days completely re-oxidizes the specimen.

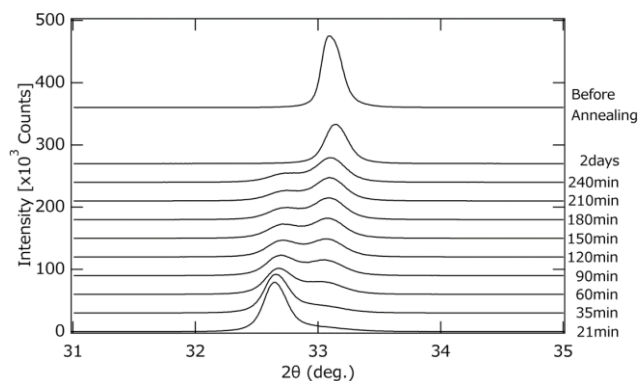


Fig. 2: Dependence of XRD spectrum of CeO₂ thin film annealed at 1273 K on exposure time to atmosphere [5].

Our previous reports show that oxygen vacancies, the accompanying Ce³⁺ state and the resulting lattice expansion induced by energetic ion irradiation are much more stable than those produced by high temperature annealing in vacuum. Even after the exposure to the atmosphere for more than a few months, we could find the lattice expansion and the ferromagnetic state of CeO₂ irradiated with energetic ions [3,4]. The high temperature annealing in vacuum produces only oxygen vacancies, but in the case of energetic ion irradiation, not only oxygen vacancies but also oxygen interstitial atoms and the disordering of Ce atom arrangements are produced. As such complicated defects prevent oxygen atoms from moving freely in the irradiated CeO₂ specimens, the oxygen vacancies in the ion-irradiated CeO₂ can survive the long-time exposure to the atmosphere.

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