

Sulfur K-edge XAFS Study of Supported Ruthenium Cluster Catalysts

Yasuo Izumi^{*1}, Atsushi Ishiguro², Taketoshi Minato¹, Takayuki Nakajima², Yasuo Wakatsuki²

¹ Interdisciplinary Graduate School of Science and Engineering, Tokyo Institute of Technology, 4259 Nagatsuta, Midori-ku, Yokohama 226-8502

² RIKEN (Institute of Physical and Chemical Research), 2-1, Hirosawa, Wako, Saitama 351-0198

Sulfur K-edge is the direct electronic and structural probe of catalytic fixation for SO₂. However, only limited studies of S K-edge were reported for heterogeneous catalysts because S K-edge is in soft X-ray region, and it is difficult to measure the edge for heterogeneous catalysts in the presence of reaction gas. Instead of S K-edge measurements in vacuum chamber, conventional fluorescence detection setup was used for supported ruthenium cluster catalysts by purging all the beam path from the XYslit in front of I₀ ion chamber to sample room with He. The I ion chamber was purged with N₂ (Figure 1). The sample catalysts were prepared by reacting [Ru₆C(CO)₁₆]²⁻ cluster crystal with TiO₂ powder. Ru content was 1.5wt%. The SO₂ fixation reactions were performed in the presence of H₂ at 443 – 503 K in gas flow system at ambient gas pressure.

The supported cluster catalysts, denoted as [Ru₆C]/TiO₂, exhibited remarkable contrast of reactivity; the conversion of SO₂ was 85 and 0% at the reaction temperatures 503 and 443 K, respectively. Figure 2 shows the S K-edge

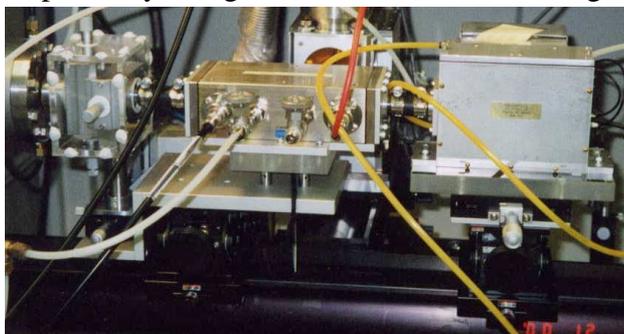


Figure 1. The detection system of S K-edge XAFS measurements utilizing He purge.

XANES spectra after the SO₂ fixation reactions for the [Ru₆C]/TiO₂ catalysts. Two strong peaks were observed at 2474.1 and 2479.0 eV in the case of reaction at 503K (Figure 2a) near the edge. The two peaks were also observed in the case of reaction at 443K, but two peaks at 2472.6 and 2482.7 eV were rather stronger (Figure 2b).

Compared to S K-edge spectrum for RuS_x, the peaks at 2472.6 and 2482.7 eV are ascribed to S²⁻ species. Taken the edge jump of Figure 2a and b (0.13 and 0.011, respectively) into account, the intensity ratio of the peak of 2472.6 eV was 17 : 4.9 for Figure 2a and b. Ionic Ru sites reacted with sulfur, e.g. RuS₂, may be more reactive for the reaction SO₂ + 2H₂ → S + 2H₂O. The assignment of the peaks at 2474.1 and 2479.0 eV are under investigation.

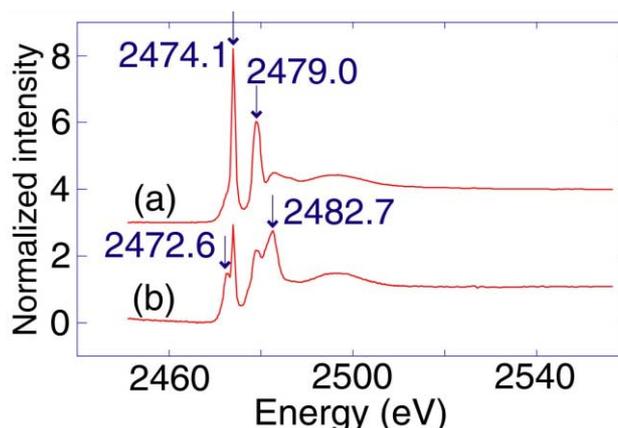


Figure 2. S K-edge XAFS spectra of [Ru₆C]/TiO₂ catalyst after the SO₂ fixation in H₂ at 503 (a) and 443K (b).

* yizumi@chemenv.titech.ac.jp