

Local Structure Analysis of Unsupported MoS₂ Based Catalysts

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Introduction

For the upgrading of heavy oils, much attention has been paid to the slurry phase hydrotreating system. Unsupported MoS₂ based catalysts are essential for this system and more improvements of catalytic activity of hydrogenation are required. In the present study, we prepared two kinds of unsupported MoS₂ catalysts by the different methods and surveyed the local structure by EXAFS for the investigation of the structure-activity relationship. In addition, the effect of Ni promoter for those catalysts was discussed.

Experimental

Two kinds of MoS₂ unsupported catalysts were prepared by the thermal decomposition of ammonium tetrathiomolybdate at 673 K. One was treated under a flow of 10% H₂S/H₂ (MS-H), and the other was prepared under 8 MPa of H₂ with vapor (MS-W). Ni promoted catalysts were prepared by the impregnation of ammonium nickel sulfate to MS-H by changing the Ni/(Ni+Mo) ratio from 0.1 to 0.9 (N-MS-H).

The catalytic activity was measured by the hydrogenation of 1-methylnaphthalene (1-MN) at 603 K. The characterization of catalysts was performed by N₂ adsorption, XRD, XPS, TEM, and Mo K-edge EXAFS. The spectrum of EXAFS was collected in the transmission mode at PF BL-10B. The catalyst was pressed into a self-supporting wafer and settled in a glass cell with Kapton windows without exposing to air.

Results and Discussion

Table 1 summarizes the hydrogenation activity and surface area of the catalysts. The rate of 1-MN for MS-W was five times higher than that for MS-H. In addition, the surface area extremely differed between two unsupported MoS₂. Fig. 1 represents the Fourier transform EXAFS of MS-H and MS-W. Both of the curves exhibited two main peaks corresponding to Mo-S (2.4 Å) and Mo-Mo scattering (3.2 Å). The local structure around Mo atom was the same as that of MoS₂ for both of the catalysts. Note that the peak of Mo-O scattering was not observed. These results were corresponded to the results of surface analysis by XPS. Therefore, two catalysts had quite similar local MoS₂ structure.

On the other hand, XRD patterns reflected the different structure of particles. Although XRD patterns of both two catalysts suggested the highly distorted MoS₂ layers, the

(002) reflection of MS-W was much smaller than that of MS-H. It indicated that the lower stacking layers of MoS₂ in MS-W. The previous researches suggested that the curvature of the basal planes and low stacking layers of MoS₂ crystallites were effective for the hydrogenation activities [1, 2]. That is, the low stacking layers with high distortion of MS-W affected the high catalytic activity and high surface area.

The effects of Ni addition to MS-H catalysts did not appear in the hydrogenation activity (Table 1). The peak of Mo-Mo scattering in Fourier transformed EXAFS increased by the addition of Ni (Fig. 2). It suggested the change of MoS₂ crystal size. In fact, TEM images exhibited the linear MoS₂ layers. The Ni addition reduced the curvature and resulted in the decrease in the activity. Above results suggested the improvement of preparation method for Ni promoted MoS₂ catalyst must be needed.

References

[1] Y. Iwata et. al., Catal. Today, 45, 353 (1998).
[2] Y. Yokoyama et. al., Am. Chem. Soc. Div. Pet. Chem., 42, 353 (1998).

Table 1 Catalytic activity and surface area

catalyst	rate constant of 1-MN (10 ⁻⁵ mol/mol-metal/s)	surface area (m ² /g)
MS-W	43	251
MS-H	7.8	9
N-MS-H ^{*1}	7.0	

^{*1} Ni/(Ni+Mo) = 0.7

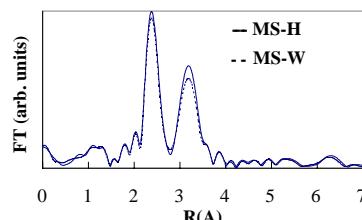


Fig. 1 Fourier transformed EXAFS of unsupported MoS₂

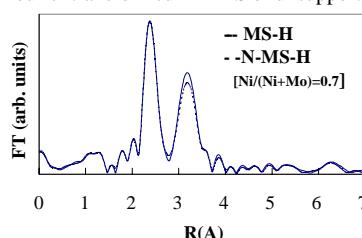


Fig. 2 Fourier transformed EXAFS of Ni promoted MoS₂