

XAFS analysis of MoO₃/MgO catalysts-effects of loading

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Introduction

A hard-to-epoxidize olefin, allyl acetate, was successfully epoxidized with t-butyl hydroperoxide over MoO₃/MgO catalysts[1]. The selectivity was dependent upon loading of MoO₃ and pore size of MgO. XAFS analysis was carried out to elucidate the structure of MoO₃ supported on MgO. We reported the relationship between the selectivity of epoxide and the pre-edge height of Mo K edge[2]. Here we'll show the effects of MoO₃ loading on the pre-edge of Mo K edge.

Experimental

A known amount of (NH₄)₆Mo₇O₂₄·4H₂O was dissolved in deionized water and MgO(Nacalai Tesque Co., particle size 25 μm) was added to the solution. After stirring for 30 min, the solution was evaporated to dryness. The solid portion was dried at 353K overnight, followed by calcination at 823K for 3 h in air. X-ray absorption spectra(Mo K-edge) were obtained at the Beam Line 10B station by a transmission mode.

Results and discussion

The XANES spectra of 2-30 wt% MoO₃/MgO, MoO₃ and MgMoO₄ are shown in Fig. 1. The intense pre-edge peaks are assigned to tetrahedral structure. The XANES spectra of 2-30 wt% MoO₃/MgO are almost overlapped with that of MgMoO₄ which is a typical tetrahedral compound. To get more precise differences the second derivatives of XANES spectra are shown in Fig. 2. Small but decisive differences were observed. The spectrum of 7 wt% MoO₃/MgO is almost identical to that of MgMoO₄. The positions of minimum points in the spectra of 2 and 5 wt% MoO₃/MgO are the same, but their depths are narrower than that of MgMoO₄. In the spectra of MoO₃/MgO with 10 wt% and above loadings the positions and depths of their minimum points are different from those of MgMoO₄. The epoxide yield of allyl acetate is highest at 7 wt% loading[1]. The results of the second derivative of the XANES spectrum and maximum epoxide yield at 7 wt% loading suggest that the catalytic active species of MoO₃/MgO for epoxidation of allyl acetate with t-butyl hydroperoxide is tetrahedral MoO₄²⁻ species supported on large pore-sized MgO[2]. In fact MgMoO₄ itself has no catalytic activity. The slight deviation from regular tetrahedral structure deteriorates selectivity in the formation of the epoxide. The XPS analysis showed that 7 wt% loading is just equivalent to the monolayer amount dispersed on the surface of MgO[1].

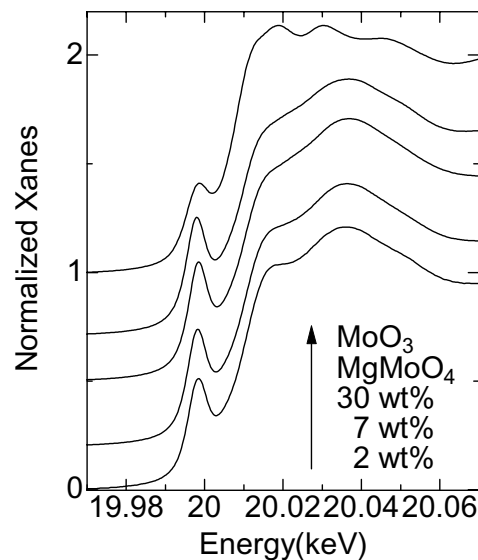


Fig. 1 Normalized XANES spectra of 2-30 wt% MoO₃/MgO and standard molybdenum oxides.

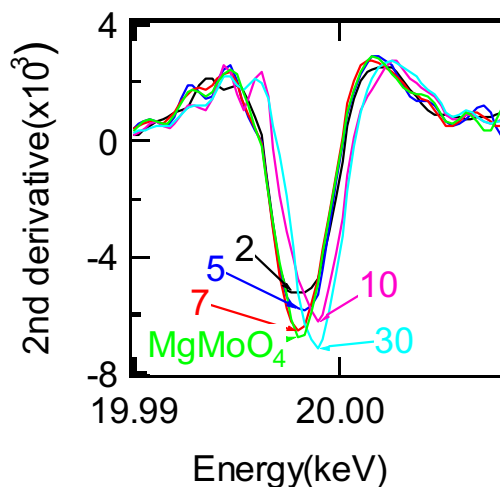


Fig. 2 The second derivatives of XANES spectra of MoO₃/MgO with various loading and MgMoO₄.

References

- [1] K. Shimura, H. Kanai, K. Utani, K. Matsuyama, S. Imamura, *Appl. Catal. A*, 283(2005) 117.
- [2] K. Shimura, T. Nishiguchi, H. Kanai, S. Imamura, *Photon Factory Activity Report 2003 #21 Part B*, 35.

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