

## Soft x-ray emission from valence band to S 2p in $\text{CuIr}_2\text{S}_4$

Yukihiro TAGUCHI\*, Katsuyuki Kitamoto, Kojiro MIMURA, Tatsunori NODA, Yukihiro FUJITA, Kouichi ICHIKAWA, Osamu AITA, Hiroki ISHIBASHI  
Osaka Prefecture Univ., Gakuencho, Nakaku, Sakai, Osaka 599-8531, Japan

Thiospinel  $\text{CuIr}_2\text{S}_4$  exhibits simultaneous structural, metal-to-insulator (MI) and paramagnetic-to-diamagnetic transitions around 230 K with decreasing temperature. The formal Ir valency is 3.5+. It has been considered that charge ordering (CO) of Ir ions takes place in the insulating phase of  $\text{CuIr}_2\text{S}_4$  [1]. The CO of Ir ions in  $\text{CuIr}_2\text{S}_4$  may partly result from electron correlation among Ir 5d electrons. Such effect has been suggested by the fact that the valence-band (VB) structure of  $\text{CuIr}_2\text{S}_4$  obtained by photoemission spectroscopy (PES) disagrees with band-structure calculations with respect to Ir 5d-derived structure despite qualitative agreement for Cu 3d- and S 3p-derived features [2,3]. It is considered that the electronic states near the Fermi level  $E_F$  or the VB maximum plays an important role in the MI and CO transitions of  $\text{CuIr}_2\text{S}_4$ . According to the band-structure calculations the states near  $E_F$  are mainly ascribed to Ir 5d-S 3p hybridized band [2]. In this work we have investigated the S 3p states hybridized with Ir 5d in  $\text{CuIr}_2\text{S}_4$  by means of soft x-ray emission spectroscopy (XES). When a sulfur 2p core-hole is created, radiative transitions from VB to S 2p reflect the S 3p hybridized with *nd* and/or *ns* orbitals because of the dipole selection rule for XES and of the localized nature of the core hole. Cu ion of  $\text{CuIr}_2\text{S}_4$  has essentially a  $3d^{10}$  configuration [3].

Measurements were carried out at beamline BL-19B. The sample used was a sintered poly-crystal. The sample surface was cleaned in situ by scraping.

Fig. 1 shows the S  $L_{2,3}$  XES spectrum of  $\text{CuIr}_2\text{S}_4$  at 275 K (open circles).  $E_F$  corresponds to the S  $2p_{1/2}$  binding

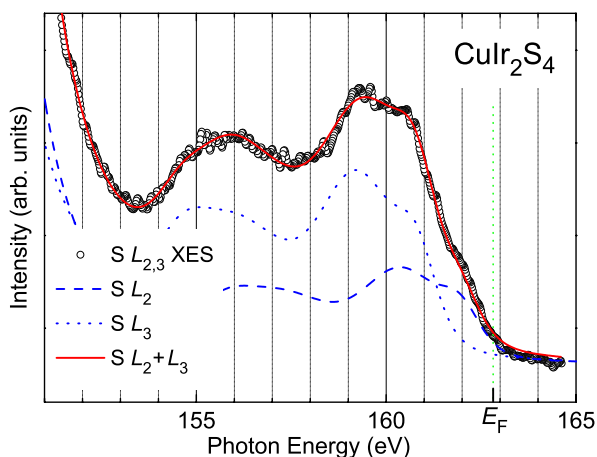


Fig.1. S  $L_{2,3}$  soft x-ray emission spectrum of  $\text{CuIr}_2\text{S}_4$  at 275 K (open circles). Broken, dotted and solid lines are decomposed  $L_2$  and  $L_3$  emission spectra and the sum of them, respectively.

energy of  $\text{CuIr}_2\text{S}_4$  determined by PES. A shoulder structure in XES is seen at 0.8 eV below  $E_F$  in agreement with the previous PES measurement (dots in Fig. 2). A previous PES study has revealed that the structure has contribution from Ir 5d [4]. Thus the structure in XES results from the S 3p-Ir 5d hybridized band in accord with the band-structure calculations. In order to confirm that the S 2p spin-orbit splitting of  $\sim 1.2$  eV does not cause the structure, we decomposed the S  $L_{2,3}$  spectrum of  $\text{CuIr}_2\text{S}_4$  into  $L_2$  ( $2p_{1/2}$ ) and  $L_3$  ( $2p_{3/2}$ ) spectra. We assumed that the transition probability from VB to S  $2p_{1/2}$  and  $2p_{3/2}$  depends only on the statistical weight of each core state, i.e. intensity ratio of 2:1 and that each spectrum can be reproduced by superposition of several Voigt functions. The obtained  $L_2$  and  $L_3$  spectra are also shown in Fig. 1. The S  $L_2$  XES, PES and theoretical spectra are compared in Fig. 2. The S  $L_2$  spectrum shows that the S 3p states hybridized with Ir 5d are distributed over almost the entire energy region of VB in good agreement with S 3p partial density of states obtained by the band-structure calculations [2].

### References

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\* taguchi@ms.osakafu-u.ac.jp

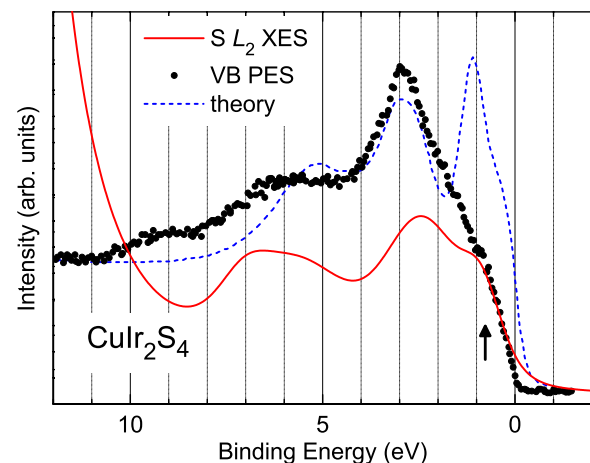


Fig. 2. Comparison of S  $L_2$  x-ray emission (solid line) with photoemission (dots) and theoretical spectra (broken line) given in Ref. 3.