

**Bulk modulus and linear compressibilities of phase G,  $\text{Mg}_{1.24}\text{Si}_{1.76}\text{H}_{2.48}\text{O}_6$** 

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X-ray diffraction data of phase G at high pressures were measured with the modified Merrill-Bassett type diamond anvil cell [1]. The data at 4.1 GPa were measured using synchrotron radiation ( $\lambda=0.6992\text{\AA}$ ) at the beam line BL-10A, Photon Factory, High Energy Accelerator Research Organization, Tukuba, Japan. The data at 5.7 and 6.8 GPa were measured using  $\text{MoK}\alpha$  radiation (50kV,40mA).

The single crystal (47x35x12 $\mu\text{m}$ ) used in this study is phase G,  $\text{Mg}_{1.24}\text{Si}_{1.76}\text{H}_{2.48}\text{O}_6$  synthesized at 1050°C and 22 GPa [2] which was previously used for the structure determination at ambient pressure (Kudoh et al., 1997)[3]. The fluid pressure medium was a 4:1 mixture of methanol:ethanol. The pressure was calibrated with the ruby fluorescence method.

Using the unit cell data (Table 1; Fig. 1) and a Birch-Murnaghan equation of state, the isothermal bulk modulus was calculated as  $K_{\text{OT}}=150(4)$  GPa assuming  $K'_{\text{OT}}=4$ . The linear compressibilities of the  $a$  and  $c$  axes are  $1.70(3)\times 10^{-3}$   $\text{GPa}^{-1}$  and  $2.56(7)\times 10^{-3}$   $\text{GPa}^{-1}$ , respectively. The  $c$ -axis is more compressible than the  $a$ -axis. The present bulk modulus value of phase G is 10% smaller than the 166 GPa value of phase D,  $\text{Mg}_{1.11}\text{Si}_{1.89}\text{H}_{2.22}\text{O}_6$  (=phase G) reported by Frost and Fei (1999)[4]. The results of structural analyses with space group  $P31m$  (No.162) at high pressures ( $R_w=6.1\%$  for 47  $F_o$  at 4.1 GPa.,  $R_w=8.0\%$  for 23  $F_o$  at 5.7 GPa and  $R_w=7.0\%$  for 18  $F_o$  at 6.8 GPa) showed that the (Si,Mg)-O distance at the S-site approaches to the Si-O distance of stishovite at corresponding pressures.

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Table 1. Crystallographic data for phase G

P (GPa)	Ambient*	4.1	5.7	6.8
$a$ ( $\text{\AA}$ )	4.790(3)	4.752(1)	4.739(1)	4.725(3)
$c$ ( $\text{\AA}$ )	4.344(3)	4.299(3)	4.284(2)	4.266(5)
$V$ ( $\text{\AA}^3$ )	86.3(2)	84.07(9)	83.32(5)	82.5(1)

\*Kudoh et al. (1997).

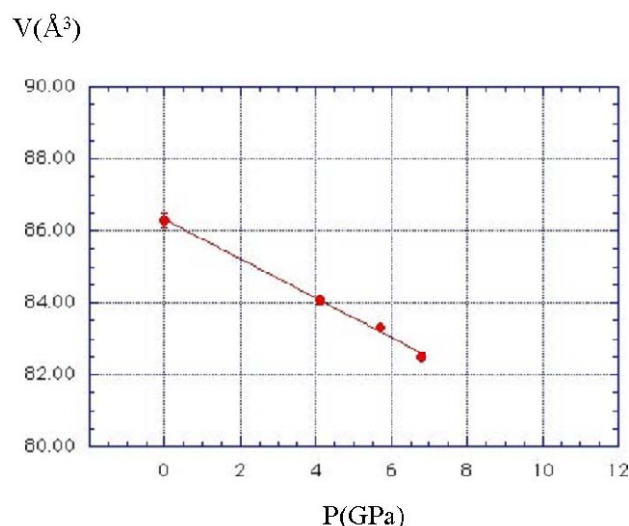


Fig. 1. Unit cell volume ( $V$ ) versus pressure ( $P$ ) for phase G. The unit cell volume at ambient pressure is from Kudoh et al. (1997)[3].

### References

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