Synthesis of Silver Particles in Tween20/Water/Ionic Liquid Microemulsions

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Introduction

Developments of room-temperature ILs as a reaction medium for inorganic nanomaterials have received much attention and could offer many opportunities and challenges for the synthesis of nanoparticles with unique shape and structures [1-3]. Although many researches have been performed focusing on the synthesis of metal nanoparticles in ILs, there are a few reports that described the reduction of metal ions and the time evolution of the formation of metal particles and aggregates in water-in-ILs microemulsions. In this study we have synthesized Ag particles by the photoreduction of silver perchlorate (AgClO₄) in the water-in-ILs microemulsions constituted Tween20/water/1-butyl-3-methylimidazolium of tetrafluoroborate ([BMIm][BF₄]) and Tween20/water/1octyl-3-methylimidazolium tetrafluoroborate ([OMIm][BF₄]) ternary systems by means of SAXS measurements.

Experimental

Colloidal dispersions of Ag particles were synthesized by the photochemical reduction of AgClO₄ in the presence of Tween 20 in water-in-[BMIm][BF₄] or waterin-[OMIm][BF₄] microemulsions. For example, 2 mL of Tween 20 was added to 4 mL of [BMIm][BF₄], followed by the addition of 5 mg benzoin and mixed vigorously. Just before the irradiation of a 500W super-high-pressure mercury lamp, 20 µL of 1.32 M AgClO₄ aqueous solution was added to the mixture solution with the simultaneous ultrasonication. Subsequently, the Ag⁺-containing waterin-[BMIm][BF,] microemulsions obtained were poured into a quartz cell, and the irradiation of UV-light was started with continuous stirring using a magnetic stirrer. In this case, the weight fraction of Tween20 was 0.33, and the $[BMIm][BF_4]$ -to-Tween20 molar ratio (R) and the water-to-Tween20 molar ratio (w) was 8.7 and 0.62, respectively. SAXS measurements were performed at BL-15A. The scattering data was collected by a position sensitive proportional counter (PSPC).

Results and Discussion

Fig. 1 shows the SAXS profiles of the colloidal dispersions of Ag particles in the (a) water-in- $[BMIm][BF_{4}]$ and water-in-[OMIm][BF₄] (b) microemulsions in the presence of Tween 20 before and after the photoirradiation. SAXS profile of the Ag⁺-

(a) [bmim][BF] [bmim][BF] + AgCIO 10 [bmim][BF] + AgCIO, + Tween20 Intensity / a.u. 0 min 60 min 120 min 10 $[Ag^{+}]=4.4 \text{ mM}$ 10 2 3 4 5 6 0 q / nm⁻¹ (b) [omim][BF₄] [omim][BF₄] + AgClO₄ 10⁵ [omim][BF₄] + AgClO₄ + Tween20 Reduction time ntensity / a.u. 0 min 60 min 10 [Ag⁺]=4.4 mM 10 2 3 5 6 4 q / nm⁻¹

Fig. 1. SAXS profiles of the colloidal dispersions of Ag particles.

containing water-in-ILs microemulsions in the absence of Tween 20 is also shown in addition to that of pure ILs without Ag⁺ ions. The following observations are worth noting. (1) A strong scattering peak is observed in the pure [OMIm][BF₄] while a monotonously increasing scattering is observed in the pure $[BMIm][BF_4]$. (2) Addition of AgClO₄ into both pure ILs induces an increase in scattering intensity at a small q-range (q < 1.0nm⁻¹), suggesting that Ag⁺ ions affect the structure of the water droplets dispersed in the ILs with keeping the ordered nanodomains of ILs. (3) Addition of Tween 20 into the Ag⁺-containing ILs makes a drastic change of the interference peak position in both ILs. The detailed analysis is in progress.

References

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