Electronic states of infinite-layer $Sr_{1-x}Eu_xFeO_{2+\delta}$ thin films studied by X-ray photoemission and absorption spectroscopies

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1 Introduction

SrFeO₂ has attracted much attention because it is an isostructural analog of an infinite-layer cuprate, SrCuO₂, which exhibits high transition-temperature superconductivity by carrier doping [1]. SrFeO₂ shows an antiferromagnetic insulator with a high Néel temperature of $T_{\rm N}$ = 473 K, and has tetragonal unit cells with a lattice constant of a = 0.3991 nm and c = 0.3475 nm [1]. We recently demonstrated that electron carriers can be doped into SrFeO₂ by Eu substituting for Sr sites in the form of epitaxial thin film [2]. The $Sr_{0.9}Eu_{0.1}FeO_{2+\delta}$ film exhibited as low as ~0.15 Ω cm at room temperature, which was approximately four orders of magnitude lower than that of undoped SrFeO₂, and showed semiconducting behavior [2]. In this study, we measured the core-level spectra of $Sr_{1-x}Eu_xFeO_{2+\delta}$ films by X-ray photoemission (XPS) and absorption (XAS) spectroscopies to investigate the valences of the Fe and Eu ions in them.

2 Experiment

Epitaxial thin films of $Sr_{1-x}Eu_xFeO_{2+\delta}$ with x = 0, 0.05, 0.10, 0.15, and 0.20 were fabricated on $SrTiO_3$ (001) substrates by combining pulsed laser deposition method with solid-phase reduction using CaH₂ at 280 °C for 24 hours. The typical thickness of the films was ~150 nm. The valences of the Fe and Eu ions were evaluated by Fe 2*p* core-level XPS spectra and XAS spectra near the Eu *M*-edge at beamline 2C of the Photon Factory, KEK. The Fermi levels of the samples were referred to that of Au foil in electrical contact with the samples. The XAS spectra were measured by the total-electron-yield method.

3 Results and Discussion

Figure 1(a) shows the Fe 2p core-level XPS spectra of the Sr_{1-x}Eu_xFeO_{2+ δ} films. Each spectrum is characterized by a Fe $2p^{1/2}$ -Fe $2p^{3/2}$ doublet and a weak satellite located at a binding energy of ~715 eV between the doublet peaks. It is known that the position of the Fe 2p satellite is very sensitive to the oxidation state of Fe. By comparing Fig. 1(a) with the Fe 2p XPS spectra of the SrFeO₂ and SrFeO_{2.5} films [3], we determined the valences of the Fe ions in all compositions of the Sr_{1-x}Eu_xFeO_{2+ δ} films to be divalent, irrespective of Eu contents. Figure 1(b) shows

the XAS spectra near the Eu *M*-edge in the $Sr_{1-x}Eu_xFeO_{2+\delta}$ films. As seen in Fig. 1(b), each spectrum has peaks at photon energies of ~1127 eV, ~1132 eV, and ~1136 eV in the Eu M_5 -edge. It also has peaks at ~1160 eV and ~1163 eV, and a shoulder at ~1158 eV in the Eu M_4 -edge. These spectra are completely similar to the calculated 3d-4fXAS spectra for Eu^{3+} ions as reported previously [4]. Therefore, the valences of Eu in the $Sr_{1-x}Eu_xFeO_{2+\delta}$ films are essentially trivalent, irrespective of Eu contents. These results imply that the electron carriers in Sr₁. $_{x}Eu_{x}FeO_{2+\delta}$ are supplied by Eu^{3+} ions substituting for Sr^{2+} sites, corresponding to the negative Hall coefficients of the $Sr_{1-x}Eu_xFeO_{2+\delta}$ films [2]. Considering charge neutrality of $Sr_{1-x}Eu_xFeO_{2+\delta}$ it is suggested that excess oxygen atoms δ are incorporated between FeO₂ sheets with increasing Eu contents.



Fig. 1. (a) Fe 2*p* XPS spectra and (b) XAS spectra near the Eu *M*-edge in the Sr_{1-x}Eu_xFeO_{2+ δ} thin films.

References

- [1] Y. Tsujimoto, et al., Nature **450**, 1062 (2007).
- [2] T. Matsuyama, et al., Appl. Phys. Express 4, 013001 (2011).
- [3] A. Chikamatsu, et al., J. Electron Spectrosc. Relat. Phenom., **184**, 547 (2012).
- [4] T. Kinoshita, et al., J. Phys. Soc. Jpn. 71, 148 (2002).
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