

Compressional behavior of hydrous SiO₂ glassSatoru Urakawa^{1,*}, Ikumi Yokota¹, Kenji Mibe², and Takumi Kikegawa³¹Department of Earth Sciences, Okayama University, Okayama 700-8530, Japan²Earthquake Research Institute, University of Tokyo, Tokyo 113-0032, Japan³Photon Factory, KEK, Ibaraki 305-0801, Japan

1 Introduction

Water changes the properties of silicate melts such as viscosity, density, and sound velocity, when it is dissolved into melts. It is important to study how water dissolves in silicate melts, and many studies have been performed. Here, we focus our attention on the water-bearing vitreous silica and we have studied the structure of hydrous SiO₂ glass by the X-ray diffraction analysis using synchrotron radiation under pressure.

2 Experiment

Hydrous SiO₂ glass was synthesized using the internally heated gas pressure vessel installed at ISEI, Okayama University. Crystalline SiO₂ enclosed in Pt capsule with 3.6 wt% D₂O had been fused at 200 MPa and 1500 C for 3h, then it was quenched into glass. Unfortunately, part of deuterium was substituted by hydrogen by exchange reaction with the gas in the pressure vessel through Pt capsule. Raman spectroscopy showed the quenched hydrous glass contained Si-OH(D) as well as H(D)₂O molecules.

Compression of hydrous SiO₂ glass was performed using the MAX80 system at AR-NE5C up to about 7 GPa. X-ray diffraction profiles were collected by energy dispersive method.

3 Results and Discussion

Interference function $Qi(Q)$ for hydrous SiO₂ glass at 0.1 MPa is almost the same as that of dry SiO₂ glass, except for the FSDP. The position of the FSDP of hydrous glass is at the higher Q side than that of dry one, indicating shrinkage of the intermediate range order by addition of water.

$Qi(Q)$ of hydrous glass changed gradually with pressure up to 7 GPa at 300 K (Fig. 1). Pressure variations in $Qi(Q)$ is the same as those of dry SiO₂ glass [1]. MD simulations showed that the Si-Si correlation changes with pressure and accounts for the variations of $Qi(Q)$. The radial distribution function $D(r)$ also shows the decrease of Si-Si distance with pressure without any change in Si-O and Si-Si distances. The FSDP of hydrous SiO₂ glass shift toward higher Q with pressure (Fig. 1), as shown in dry one [2]. Thus, in the hydrous SiO₂ glass, the network topology of SiO₄ tetrahedra changes with pressure.

Although hydrous SiO₂ glass takes a similar short range structure to dry one, comprising a SiO₄ tetrahedron, the medium range order of hydrous glass is smaller than

dry glass. Unfortunately, we could not find how water works in structure changes of the hydrous SiO₂ glass under pressure. Compressional behavior of hydrous SiO₂ glass agrees with dry glass; the intermediate range order shrinks with increasing pressure.

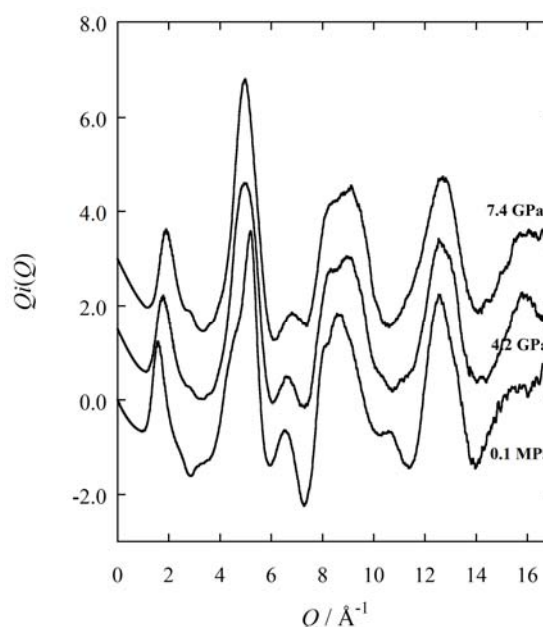


Fig. 1: Variation of interference function $Qi(Q)$ for hydrous SiO₂ glass during compression.

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References

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