

## XAFS Characterization of Molecularly Imprinted Pd-Complex Catalysts for Suzuki Cross Coupling Reaction

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### Introduction

Molecularly imprinted Pd-complex catalysts were prepared on a SiO<sub>2</sub> surface by molecular imprinting of a SiO<sub>2</sub>-supported Pd complex using SiO<sub>2</sub>-matrix overlayers. The local coordination structures of the prepared molecularly imprinted Pd-complex catalysts, which exhibited fine shape selectivity for Suzuki cross coupling reactions, were investigated by Pd K-edge XAFS.

### Experimental

A Pd precursor complex together with a template molecule, (dppe)Pd(temp) (**1**) (dppe = 1,2-ethanediybis[bis[4-(1-methylethenyl)phenyl]-phosphine], temp = 5-(2-propen-1-ol)-[1,1'-biphenyl]-2,2'-diol) was attached on a SiO<sub>2</sub> surface functionalized with *p*-styryl moiety to prepare a SiO<sub>2</sub>-supported Pd complex (**2**). SiO<sub>2</sub>-matrix overlayers were stacked on the surface of **2** by chemical vapour deposition to prepare a molecularly-imprinted Pd complex on the SiO<sub>2</sub> surface (**3**).

Pd K-edge XAFS was measured in a transmission mode at 20 K at the NW10A station with a Si(311) double-crystal monochromator. Ionization chambers filled with pure Ar and pure Kr were used to monitor incident and transmitted X-rays, respectively. EXAFS spectra were analysed using ATHENA and ARTEMIS programs. *k*<sup>3</sup>-Weighted EXAFS oscillations were Fourier transformed into *R*-space, and curve-fitting analysis was performed in *R*-space with coordination number (CN), interatomic distance (*R*), Debye-Waller factor ( $\sigma^2$ ), and correction-of-edge energy ( $\Delta E_0$ ). Phase shifts and backscattering amplitudes were calculated by the FEFF8 code.

### Results and Discussion

The local coordination structures of the supported Pd complexes were examined by Pd K-edge EXAFS analysis, whose structural parameters are listed in Table 1. The (dppe)Pd(temp) complex precursor (**1**) with the template ligand had Pd-P bond at 0.223 ± 0.001 nm and Pd-O bond at 0.205 ± 0.001 nm, whose coordination numbers (CNs) were both fixed to be 2.0.

The bond distances of the Pd-P and Pd-O coordination in **2** were fitted as 0.223 ± 0.001 and 0.206 ± 0.002 nm, respectively. The CNs of the Pd-P and Pd-O bonds were fitted as 1.7 ± 0.1 and 2.0 ± 0.4, respectively. These results suggest that the local coordination structure of **1** with the template and the dppe ligand was maintained

after it was attached to the SiO<sub>2</sub> surface together with the results of <sup>13</sup>C NMR and UV/vis spectra.

The bond distances of the Pd-P and Pd-O coordination in **3** were fitted as 0.225 ± 0.001 and 0.203 ± 0.002 nm, respectively. The CNs of the Pd-P and Pd-O bonds were fitted as 1.6 ± 0.2 and 1.9 ± 0.4, respectively. These results also indicated the retention of the local coordination structure of the Pd complex in **3** after stacking of SiO<sub>2</sub>-matrix overlayers. <sup>13</sup>C NMR and UV/vis spectra of **3** supported this result. Pd K-edge XANES of **1**, **2**, and **3** were almost identical, also supporting the structural and oxidation state similarities.

The SiO<sub>2</sub>-supported Pd complex showed significant activity for the C-C coupling reaction, and different catalytic performances were observed for various combinations of aryl iodides and arylboronic acids on the molecularly-imprinted Pd complex catalyst **3**.

**Table 1** Structural Parameters Determined by Curve-Fitting Analysis of Pd K-edge EXAFS Measured at 20 K

Shell	CN	<i>R</i> /nm	$\Delta E_0$ /eV	$\sigma^2 / \times 10^3 \text{ nm}^2$
<b>1</b> <sup>a</sup>				
Pd-P	2.0	0.223 ± 0.001	5.4 ± 0.9	2.4 ± 0.2
Pd-O	2.0	0.205 ± 0.001	6.2 ± 2.2	5.8 ± 0.8
<b>2</b> (Pd: 0.6 wt%) <sup>b</sup>				
Pd-P	1.7 ± 0.1	0.223 ± 0.001	5.4	2.4
Pd-O	2.0 ± 0.4	0.206 ± 0.002	6.2	5.8
<b>3</b> (Pd: 0.35 wt%) <sup>c</sup>				
Pd-P	1.6 ± 0.2	0.225 ± 0.001	5.4	2.4
Pd-O	1.9 ± 0.4	0.203 ± 0.002	6.2	5.8

<sup>a</sup> *k* = 30-160 nm<sup>-1</sup>, *R* = 0.12-0.22 nm, *R*<sub>i</sub> = 0.47%.

<sup>b</sup> *k* = 30-160 nm<sup>-1</sup>, *R* = 0.12-0.22 nm, *R*<sub>i</sub> = 1.0%.

<sup>c</sup> *k* = 30-140 nm<sup>-1</sup>, *R* = 0.12-0.22 nm, *R*<sub>i</sub> = 1.0%.

### Reference

[1] N. Maity, S. Muratsugu, M. Tada, submitted.

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