## Structure of (Fe,Ni)<sub>3</sub>S<sub>2</sub> under pressure

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## 1 Introduction

The Earth and planetary cores are composed of ironnickel alloy and some light elements. Sulfur is a primary candidate of lightening elements alloyed with iron-nickel. Therefore, phase relations and structure of the iron-nickel sulfides at high-pressures and temperatures are fundamental to study formation process and evolution of the planetary cores. Here we focus attention on ironnickel sulfide (Fe,Ni)<sub>3</sub>S<sub>2</sub>. Fe<sub>3</sub>S<sub>2</sub> is known to be stable above 14 GPa and affect the melting relations in the system Fe-FeS, but its crystal structure is still unknown [1]. On the other hand,  $Ni_3S_2$  is stable even at 1 atm, which has trigonal symmetry with space group R32 ( $\alpha$ -Ni<sub>3</sub>S<sub>2</sub> heazlewoodite) [2]. This phase is known to transform into the high pressure orthorhombic structure with Cmcm symmetry at high pressures [3]. In this study we examine stability field of high-pressure phase of Ni<sub>3</sub>S<sub>2</sub> and alloying effect of iron to its crystal structure.

## 2 Experiment

High pressure and temperature experiments were conducted at the pressures of 13 to 18 GPa and the temperatures up to 900 K using the MAX III system installed at PF-AR NE7. X-ray diffraction patterns were taken by an energy dispersive method using a Ge-SSD at Bragg angle of 6 degree. Pressure was evaluated by the unit cell volume of NaCl pressure marker. X-ray profiles were analyzed using the PD Indexer software package provided by Seto Y.

## 3 Results and Discussion

We observed the phase transformation from  $\alpha$ -Ni<sub>3</sub>S<sub>2</sub> to the orthorhombic structured-Ni<sub>3</sub>S<sub>2</sub> between 300 K and 500 K around 15 GPa (Fig. 1). This is close to the phase boundary determined by the first-principle calculation [4]. It is, therefore, though to be reasonable that the highpressure phase of Ni<sub>3</sub>S<sub>2</sub> is stable above 15 GPa. The orthorhombic phase can be quenched to room temperature at high pressures, but it backs into  $\alpha$ -Ni<sub>3</sub>S<sub>2</sub> after decompression. We also observed (Fe<sub>0.5</sub>Ni<sub>0.5</sub>)<sub>3</sub>S<sub>2</sub> under pressure and found the orthorhombic structure stabilized above 700 K around 15 GPa (Fig. 2). Thus about 50 % Fe<sub>3</sub>S<sub>2</sub> can dissolve into the high pressure Ni<sub>3</sub>S<sub>2</sub> phase. This implies that Fe<sub>3</sub>S<sub>2</sub> which is stable above 14 GPa may also take the orthorhombic structure with Cmcm symmetry.



Fig. 1: The *P*-*T* path of in-situ observation of  $Ni_3S_2$ . Solid squares are the orthorhombic phase and open symbols are  $\alpha$ -Ni<sub>3</sub>S<sub>2</sub>. Dashed line is a phase boundary evaluated by the first-principle calculation [4].



Fig. 2: Energy-dispersive X-ray diffraction profiles of  $(Fe_{0.5}Ni_{0.5})_3S_2$ .

<u>References</u>

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