

Small-Angle X-ray Scattering Measurement of Ionic Liquid-CO₂ Systems

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1 Introduction

Despite being salts, ionic liquids (ILs) that show liquid phase at room temperature have high CO₂ solubility and selectivity compared with gases included in the exhaust gas such as N₂, O₂, and H₂. Fluctuations, inhomogeneity of molecular distributions of the systems, provide mesoscopic structural information and are estimated by small-angle scattering methods for X-rays and neutrons. Containing heavy atoms such as fluorine, sulfur, some ionic liquids have higher X-ray absorbing power than some molecular liquids. Furthermore, in order to perform the measurement of IL-CO₂ systems at phase equilibrium without agitation, shorter exposure time is needed, which requires much higher intensity of X-ray source to perform scattering experiment.

2 Experiment

A high-pressure sample holder made entirely of titanium and a titanium alloy was designed to vary the path length precisely and perform stable operation at high temperature and pressure, while retaining a constant path length[1]. Imidazolium-based ionic liquid, 1-ethyl-3-methylimidazolium bis(trifluoromethylsulfonyl) amide [C₂mim][NTf₂] was used as a sample. The measurement for IL-CO₂ mixtures was carried out at 314 K with pressures of 0.1, 1, 5, 10, 15, and 20 MPa under the process of pressure application using the apparatus at the BL-6A. Also, using samples which has different alkyl-chain length ([C_nmim][NTf₂] (n = 2, 4, 6, 8)) at ambient condition, operation was performed.

An X-ray beam was monochromatized to $\lambda = 1.50 \text{ \AA}$ and the observable s -region was $0.02 \text{ \AA}^{-1} \sim 0.20 \text{ \AA}^{-1}$, where the scattering parameter, s , is defined as $4\pi\sin\theta / \lambda$ (2θ : scattering angle, λ : wavelength). $I(0)$, scattering intensities at $s = 0$, directly relate to the mesoscale structural fluctuation, are evaluated from the obtained SAXS intensities.

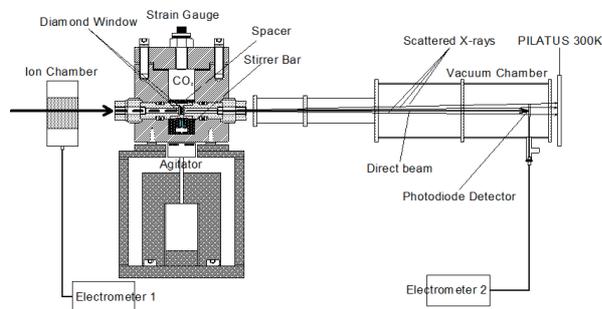


Fig. 1: Experimental layout of the SAXS measurements

3 Results and Discussion

Fig.2 and 3 shows the intensities of [C₂mim][NTf₂]-CO₂ system and of [C_nmim][NTf₂], respectively. $I(0)$,

shown in Fig.4, was obtained from the extrapolation of the each plot. At the region over 10 MPa, $I(0)$ got two or three times as $I(0)$ under 10 MPa, implying the structural change of IL at higher pressure region. And as alkyl-chain becomes longer, $I(0)$ at ambient condition showed higher value.

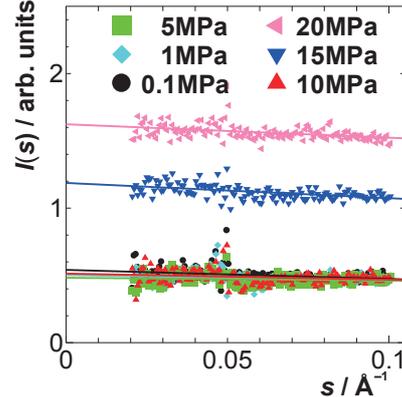


Fig.2: SAXS intensities for [C₂mim][NTf₂]-CO₂ system

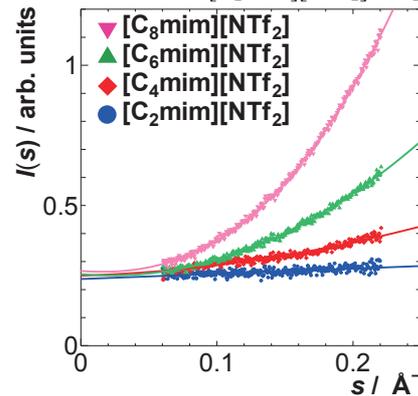


Fig.3: SAXS intensities for [C_nmim][NTf₂] at ambient condition

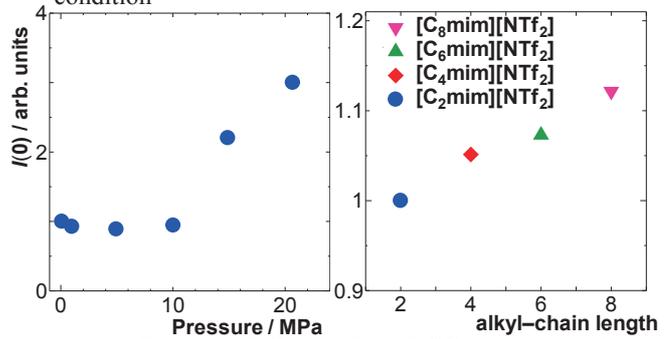


Fig.4: $I(0)$ plots of [C₂mim][NTf₂]-CO₂ system(Left) and [C_nmim][NTf₂] at ambient condition(Right)

References

[1] T. Morita, *et al.*, *Japanese Journal of Applied Physics*, **51**, 076703 (2012).

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