

## Atomic Arrangement and Electronic Structure of $\text{Al}_{72}\text{Co}_{16}\text{Ni}_{12}$ Quasicrystal Studied by X-ray Emission Spectroscopy and Cluster Calculation

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### Introduction

A decagonal quasicrystal Al-Co-Ni shows unique two-dimensional quasiperiodic atomic arrangement and associated unusual electric properties [1]. The atomic arrangement is usually regarded as a special tiling of two unit tiles, fat and skinny rhombi, with 5-fold rotational but no translational symmetry [1, 2]. It may be also viewed as building from a single unit cluster with overlap rules [3]. A band structure calculation for model approximants shows that the energy distribution of their transition-metal 3d states depends considerably on the transition-metal arrangement [4]. Thus, we have studied the electronic structure and atomic arrangement of the Al-Co-Ni quasicrystal by the Co and Ni  $L\alpha$  x-ray emission (XES) measurement and DV- $X\alpha$  cluster calculation, in order to clarify the origin of the unique structure and physical properties.

### Experimental and Calculating Procedures

A single decagonal quasicrystal  $\text{Al}_{72}\text{Co}_{16}\text{Ni}_{12}$  was used for the XES measurement after polishing its surface. The XES spectra were measured at room temperature at BL-2C of Photon Factory. Binding energies  $E_B$  of the relevant core levels were obtained by hard x-ray photoelectron measurement at BL47XU of SPring-8.

The DV- $X\alpha$  calculation was performed by a code SCAT [5] for a model cluster, which is based on a recent model proposed by the electron microscopy [2], specifying the Co and Ni sites. An example of the cluster is illustrated in the inset of Fig.1. The cluster has a triple-layered structure of two unit layers A and B with hydrogen atoms at their layer edges in the stacking sequence, for example, of A-B-A, which may be regarded as a 2 nm unit cluster of the decagonal quasicrystal [2].

### Results and Discussion

Figure 1 shows the Co and Ni  $L\alpha$  XES spectra of  $\text{Al}_{72}\text{Ni}_{12}\text{Co}_{16}$  recorded at the excitation photon energy  $h\nu$  of 841 and 923 eV, respectively, in comparison to a valence-band x-ray photoelectron spectrum (XPS) taken at  $h\nu = 852$  eV and a calculated density of states (DOS) for the model cluster. Here, the DOS is the sum of the DOS of the central layers A and B in the B-A-B and A-B-A stacking sequences, respectively, in order to reduce the surface effects. The XES spectra clearly show that a single-peaked band observed at  $E_B = 2$  eV in XPS consists of the Ni and Co 3d states and that the Ni 3d states are located lower in energy than the Co 3d ones, which is inconsistent with the prediction by the band

structure calculation for the approximant [4].

As seen in Fig.1, the calculated DOS agrees fairly well with the XES results, although the Co 3d states are rather bimodal. The transition-metal 3d bands and a pseudogap *i.e.* the reduction of the Al partial DOS, across the Fermi energy are located lower in energy as a whole than calculated for previous small model clusters [6]. The calculated Co 3d bands, however, are still incompatible with the experimental results. The Co 3d bands at the high and low binding energies are ascribed to Co atoms in the second-inner and peripheral part, respectively, of the model cluster. The previous study suggests rather random occupation of the transition-metal sites by Co and Ni [6]. Thus, further detailed study on the arrangement, including the improvement of the peripheral part of the model cluster, and chemical interaction of the transition-metals is now in progress and will be reported elsewhere.

### References

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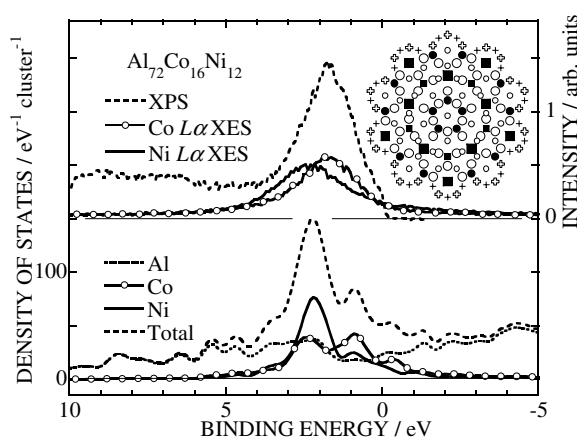


Fig.1 Co and Ni  $L\alpha$  XES spectra of  $\text{Al}_{72}\text{Co}_{16}\text{Ni}_{12}$ . A valence-band XPS spectra and calculated density of states are shown for comparison. A model cluster for the calculation is shown in the inset, where large and small symbols represent atoms in the A and B layers of the cluster, respectively: Al by open circles, Ni by closed circles, Co by squares, and H by crosses.