## **XPS and NEXAFS studies of C**<sub>60</sub>-Co films with giant tunnel magnetoresistance

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In recent years, investigations on molecular spin electronics which control a transport of spin-polarized carrier through organic molecules including C60, carbon nanotubes (CNTs) and graphene have been performed intensively.[1] Very recently, we have found that C<sub>60</sub>-Co mixture films where Co nanoparticles are dispersed into a matrix of the C60-Co compound show the large tunnel MR effect of 80-90% at low temperature.[2,3] The observed MR ratios cannot be explained by the spin polarization of crystalline Co (~40%), and hence it is expected that the electronic and magnetic structures of the C60-Co compound play a significant role to the spin-dependent transport in the C60-Co films. In the present work, we studied about local electronic states of the C60-Co films by X-ray photoelectron spectroscopy (XPS) and near-edge X-ray absorption fine structure (NEXAFS) spectroscopy. [4]

C60-Co mixture films were prepared by the codeposition method under the UHV condition  $(10^{7}Pa)$ . The experimental set-up for co-deposition consists of a Knudsencell for  $C_{60}$  (99.99%, sublimated) and an electron beam evaporator for Co (99.99%). The  $C_{60}$ -Co films with compositions of  $C_{60}Co_x$  (x: the number of Co atoms per a  $C_{60}$ molecule), and thickness of 50 nm were deposited on Ag films on MgO (001). A pure  $C_{60}$  film was prepared in a similar way. The prepared samples were transferred to an experimental chamber without breaking the UHV condition. XPS measurements were performed at Photon Factory BL-27A of High Energy Accelerator Research Organization (KEK-PF) employing Mg K<sub> $\alpha$ </sub> (hv=1253.6eV) and Y M<sub> $\zeta$ </sub> (hv=132.3eV) X-ray sources. NEXAFS measurements were performed at BL-7A of KEK-PF. The NEXAFS spectra were obtained by the total electron yield method.

Figure 1 and 2 show the XPS spectra in the valence region and the C1s NEXAFS spectra for the  $C_{60}$  film and the  $C_{60}$ -Co film ( $C_{60}Co_3$ ) composed of the  $C_{60}$ -Co compound.[5] In Figure 1, five peak components ( $A_1$ - $A_5$ ) associated with the molecular orbitals (MOs) of  $C_{60}$  are confirmed in the  $C_{60}$ film. In the  $C_{60}$ -Co films, the  $A_3$ - $A_5$  peaks lying in the range of  $E_B = 5$ -9eV show comparable shifts from the positions in the  $C_{60}$  film. Meanwhile, the structure close to the Fermi level ( $E_B$ <5eV) shows significant differences from the  $C_{60}$ film, as represented by the peak components of  $B_1$ - $B_5$ . The  $B_2$ and  $B_4$  peaks seem to be corresponding to the  $A_1$  and  $A_2$ peaks from the shifted quantities. Investigations on the  $C_{60}$ -Ti and  $C_{60}$ -Rb composites [6,7] have pointed out that, when metal atoms form covalent bonds with carbon atoms, new peaks appear close to the Fermi level ( $E_{B}$ ~3-5eV) or close to the HOMO level of C<sub>60</sub> (A<sub>1</sub>). Judging from these reported results, the observed peaks of B<sub>1</sub>, B<sub>3</sub> and B<sub>5</sub> in the C<sub>60</sub>-Co film are reasonably ascribed to the hybridized states between the C<sub>60</sub>  $\pi$  and Co 3d orbitals. The comparable shifts of the A<sub>3</sub>-A<sub>5</sub> peaks in the C<sub>60</sub>-Co film indicate the cage distortion of C<sub>60</sub>, possibly by the hybridized bond formation. The  $\pi$ -d hybridization is confirmed in the absorption spectra. In Figure 2, the first peak of the C<sub>60</sub>-Co film is shifted toward the higher energy side and the peak width is wider than the LUMO peak of the C<sub>60</sub> film. This peak broadening and shift indicate the formation of the covalent bonds with the charge transfer (back donation) between C<sub>60</sub>  $\pi$  and Co 3d orbitals.



Figure 1 XPS spectra of the  $C_{\scriptscriptstyle 60}$  and  $C_{\scriptscriptstyle 60}\text{-}Co$  films  $(C_{\scriptscriptstyle 60}\text{Co}_{\scriptscriptstyle 3})$  in the valence region



Figure 2 NEXAFS spectra of the  $C_{60}$  and  $C_{60}$ -Co films  $(C_{60}Co_3)$  by C1s core excitation.

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