

## Nanophase-separated structures of AB block copolymer / C homopolymer blends with hydrogen bonding interaction

Atsushi Takano, Katarzyna Dobrosielska, Sou Wakao, Daisuke Kawaguchi and Yushu Matsushita  
Department of Applied Chemistry, Nagoya University, Furo-cho, Nagoya 464-8603, Japan

### Introduction

The block copolymers consisting of incompatible polymers form self-assembled structures in bulk[1], such as spherical-, cylindrical-, bicontinuous- and lamellar ones depending on their compositions. Polymer blends of AB/B type also show the same morphological change with composition, however amounts of the homopolymer solubilized into corresponding domains of AB block copolymer are limited. [2] In this study morphology of AB/C polymer blends with attractive interactions such as hydrogen bonding interaction was investigated by transmission electron microscopy (TEM) and small-angle X-ray scattering (SAXS).

### Experimental

poly(styrene-*b*-2-vinylpyridine) diblock copolymer (SP,  $M_w=130K$ , weight fraction of polystyrene  $w_s=0.67$ ), poly(2-vinylpyridine) (P,  $M_w=3K$ ), and poly(4-hydroxystyrene) (H,  $M_w=8k$ ) were used as component polymers. Blend sample films of SP/P and SP/H were prepared by solvent casting from THF solutions and annealing at  $150^\circ\text{C}$  for 5 days under vacuum.

### Results&Discussion

The morphology of parent SP diblock copolymer shows hexagonally-packed cylindrical structure. As P was added to the SP the blend's morphology transferred to lamellar ones. However addition of more P homopolymer causes macrophase separation, as shown in Figure 1. In contrast, in SP/H blends, macrophase separation of their

components did not exist and various nanophase separated structures were observed in wide composition range as shown in Figure 1 and 2. It is due to miscibility of P and H via hydrogen bonding interactions, which are considered as the driving force producing uniform nanophase separated structures in wide composition range.

### References

- [1] Y. Matsushita *J. Polym. Sci. Part B: Polym. Phys.* 38, 1645 (2000).  
[2] T. Hashimoto et al. *Macromolecules*, 24, 240 (1991)

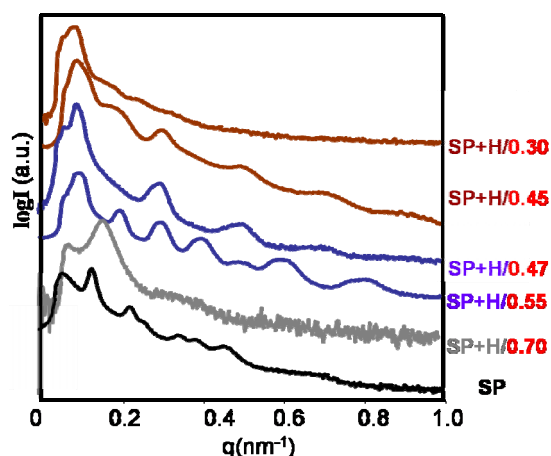


Figure 2. SAXS patterns of SP/H blends

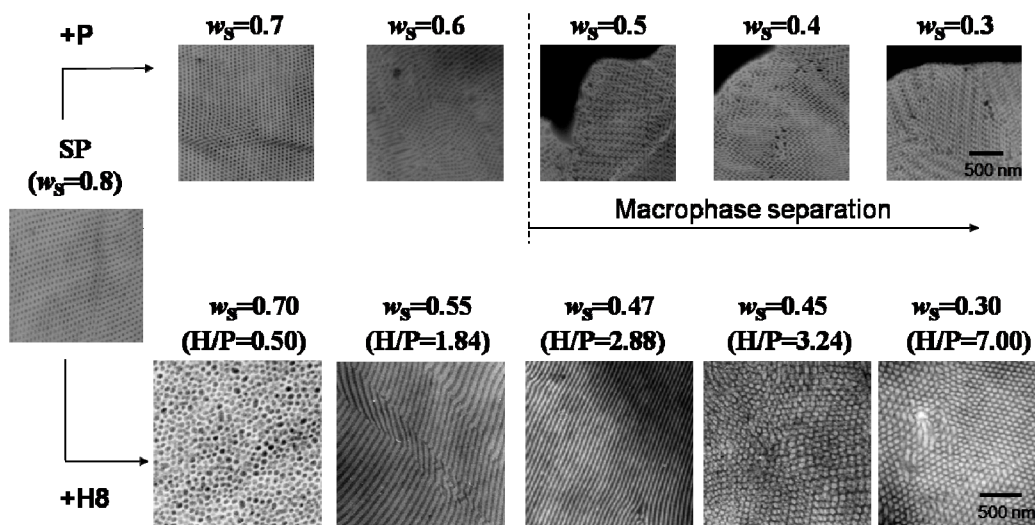


Figure 1. TEM images of SP/P blends and SP/H blends